PHYSICO-CHEMICAL PROPERTIES OF WATER OF RAMSHAGAR DIGHI, THE LARGEST MAN MADE HISTORICAL RESERVOIR IN NORTHERN BANGLADESH

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Abstract

Ramshagar dighi is a larger historical man made reservoir (Dighi, Bengali meaning) situated at Tajpur village in Dinajpur, Bangladesh. This study was aimed to estimate current status of physico-chemical variables of water of Ramshagar dighi at Dinajpur District, Bangladesh. Monthly average changes in physico-chemical parameters such as water temperature, total dissolved oxygen, air temperature, humidity, rainfall, water depth and pH of water were analyzed for the period of 11 months from May 2011 to March 2012. The average air temperature (°C) at the study area of Ramshagar dighi at Dinajpur District was determined as 24.97 ± 4.92. Our present study showed that physico-chemical properties of water in Ramshagar dighi were monthly average of water temperature (°C) as 24.68 ± 4.77, air temperature (°C) 24.97 ± 4.92, humidity as 82.075 ± 4.14, rainfall as 1534.5 mm, water depth as 9.10 m ± 1.286, pH as 7.67 ± 0.48 and carbon dioxide as 0.85 ± 0.92 as well as dissolved oxygen as 4.65 ± 0.62 respectively during the period of May 2011 to March 2012. Therefore, present study was conducted to assess physico-chemical properties of water of Ramshagar dighi, Dinajpur, Bangladesh.

Key words: Northern Bangladesh, physico-chemical properties, Ramshagar, reservoir, water

Introduction

Bangladesh is a fertile land for aquaculture. It has a wide variety of dynamic ecosystem, viz. rivers, canals, mangrove forests, natural lakes, man-made reservoirs, freshwater marshes, oxbow lakes, freshwater depressions and seasonally inundated extensive floodplains (Akonda 1989, IUCN 1993). Rivers, ponds and lakes are the waterways of strategic importance across the world, providing main water resources for domestic, industrial and agricultural purposes (Faith 2006). Water is essential for the survival of any form of life. Of the total water present on earth, only 33,400 m³ are available for drinking, agriculture, domestic and industrial consumption (Dara 2007). Surface waters are vital and vulnerable freshwater resources that are critical for the sustenance of all life. Water quality parameters are the crucial elements for aquaculture. These water reservoirs that contained water are the main sources of fisheries production. A successful aquaculture is completely dependent on the water quality parameters that arise from a magnitude of physical, chemical and biological interactions. The physico-chemical characteristics of the aquatic system have a direct influence on the types and distribution of aquatic biota (De 2007). Water quality is patho-physiological condition of fish. It is not only the suitability of water for the survival and growth of fish but also is the indicator of aquatic pollution and diseases in fish which is normally governed by only a few variables. Patho-physiological condition of fish depended on the physico-chemical properties of water. Fishes are more dependent on water temperature, pH, dissolved oxygen, free carbon dioxide, alkalinity and some other salts

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for growth and development (Nikolsky 1963). Mollah and Haque (1978) reported that the physico-chemical factors of water and soil have some effects on plankton periodicity. The study of physico-chemical properties included the physical, chemical and biological parameters of a water body and these are interrelated and have direct effect on the productivity of a water body (Welch 1952). These physico-chemical parameters refer to the temperature, turbidity, odour, colour, total solid, total dissolved solid, total suspended solid, pH, conductivity, iron content, acidity, total hardness, and chloride content (FAO 1984). The quality of water in any ecosystem provided significant information about the available resources for supporting life in that ecosystem. The key feature of an ecosystem is the interaction between the biotic and abiotic components. Good quality of water resources depended on a large number of physico-chemical parameters and biological characteristics. All the vital functions of fish like feeding, digestion, assimilation, growth, response to stimuli and reproduction are depended on water quality. Thus healthy aquatic ecosystem is depended on the physico-chemical and biological characteristics (Venkatesharaju et al. 2010).

Nowadays the physico-chemical properties of water is altered due to the accumulation of large quantities of hazardous contaminants such as heavy metals and organic micro pollutants in the sediments of lakes, rivers and marine areas world-wide (Tuncer et al. 1993). Ramshagar dighi as a pond is highest and largest historical man made pond in the district of Dinajpur, Bangladesh. It bears some exceptional historical facts of natural heritage. As a historical reservoir the significance of the study is the demand of time. Therefore the objective of present study is to assess the physico-chemical properties of water of Ramshagar dighi. Because there is a very little work has been done on the study of physico-chemical properties of water of Ramshagar dighi, Dinajpur, Bangladesh. So it is important to study the physico-chemical properties of water of Ramshagar dighi for the sake of profitable aquaculture and for the upliftment of socio-economic condition of the general people.

Materials and Methods

The present investigation was conducted through a period of 11 months from May 2011 to March 2012 in the study area of Ramshagar dighi, Dinajpur, Bangladesh. The sampling and physical test was done monthly during the study period.

Meteorological data: During this study period meteorological data of the study area were collected from meteorological regional station, Dinajpur.

Air and water temperature: Air temperature of the study area was taken from Meteorological Regional Station Dinajpur. Water temperature was taken by using a centigrade mercury thermometer with a range of 0° to 120°C at the time of sampling. The bottom temperature of water was recorded by descending the thermometer until reaching to the bottom.

Water quality measurement: Sampling at the study area was carried out fortnightly from May 2011 to March 2012. Water samples were collected from the depth of 20 - 30 cm below the surface and also from the bottom. Physical data of the study area were recorded immediately.
Physico-Chemical Properties

**Depth of water:** Water depth of the study area at the time of sampling was noted monthly. Depth of water was measured by the help of a meter scale with the ranges of 0 - 20 m. The depth was also measured by a graduated rope at various places of the study area.

**pH (potenz hydrogen):** pH refers to the amount of hydrogen ions in a solution. The pH value of water was determined by using a digital pH meter (Model HI, Hanna 96107).

**Free Carbon dioxide (fCO₂):** Free CO₂ (mg/l) was determined by titration of the water sample with NaOH solution (sodium hydroxide) using Phenolphthalein as an indicator (Welch 1948). The test was done immediately after collecting the sample.

**Dissolved oxygen (DO):** The dissolved oxygen (mg/l) was estimated by using the Winkler’s Method (APHA 1976). Manganese sulphate, sulfuric acid, starch solution and sodium thiosulfate were used as reagents for the determination of dissolved oxygen. A sample of 300 ml glass biological oxygen demand (BOD) stopper bottle brim was filled with sample water which carefully added to 2 ml of manganese sulphate so that no bubble can be introduced. Alkali-iodide-azide and 2 ml of concentrated sulphuric acid were added respectively. If oxygen is present, a brownish-orange cloud of precipitate or flock will appear where the floe was settled down by turning the upside down at several times. Carefully stopper and invert several times to dissolve the floe. At this point, the sample was fixed and stored for 8 hours in a cool and dark place by the lid off with aluminum foil and rubber band. The sample was titrated with sodium thiosulfate up to pale straw color. Next 2 ml of starch solution was added, which imparted blue color to the solution. Thus the titration indicated the culminating point of the experiment and the data were recorded.

**Results and Discussion**

Mean values of the water parameters such as air temperature, humidity, rainfall, water depth, water temperature, total dissolved oxygen, free carbon dioxide and pH in Ramshagar dighi, Dinajpur, Bangladesh were represented below.

**Air temperature:** Data regarding the air temperature at the study area during May 2011 to March 2012 was observed as a monthly average maximum and minimum variation of temperature of 33.03°C to 10.88°C in June 2011 and January 2012 (Table 1 and Fig. 1) respectively throughout the study period. Monthly average variation of maximum air temperature (°C) was recorded as 29.63 ± 3.93 and minimum was recorded as 20.30 ± 6.03. Air temperature showed fluctuation as lower in winter and higher in summer, spring and autumn. Thus the impact of air temperature on the fluctuation of water temperature was reported by Ismail et al. (1984) and Begum et al. (1989) in Bangladesh. In pond, bottom water temperature is slightly lower than or equal to surface water temperature (Rahman et al. 1982). Air temperature may fluctuate in the study area due to solar radiation, season, length of the day, geographical position and other meteorological conditions, which were not considered in this study.
Table 1. Monthly average fluctuation of air temperature, humidity, water depth, rainfall, water temperature, pH, free carbon dioxide and total dissolved oxygen in Ramshagar dighi, Dinajpur, Bangladesh from May 2011 to March 2012.

<table>
<thead>
<tr>
<th>Parameters</th>
<th>Months</th>
<th>2011 Mn ± SD</th>
<th>2012 Mn ± SD</th>
</tr>
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<td></td>
<td>Max</td>
<td>Min</td>
<td>Av</td>
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<td>5.8</td>
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</tbody>
</table>

AT = air temperature, HD = humidity, WD = water depth, RF = rainfall, WT = water temperature, pH = potenz hydrogen, fCO = free carbon dioxide, DO = dissolved oxygen, Max = maximum, Min = minimum, Up = upper, Md = middle, Lw = lower and Av = average.

**Humidity:** During the study period Relative Humidity was recorded throughout the study period. The maximum humidity of 97.37% was recorded in December 2011 and the minimum one was 55.63% in February, 2012 (Table 1 and Fig. 1). The maximum relative humidity was determined as 94.73 ± 1.51 and minimum relative humidity was as 69.42 ± 7.26. Monthly average relative humidity was recorded as 82.075 ± 4.14. This study is supported by a similar observation made by Islam and Mendes (1976).

**Rainfall:** The rainfall during the study period in Ramshagar dighi showed a distinct seasonal trend of fluctuation. It was recorded maximum as 383.3 mm in August 2011 and the minimum was as 1.2 mm in October 2011 and no rain fall during the month of December 2011 (Table 1 and Fig. 1). The average rainfall over the 11 months was found as 139.5 mm. Rainfall had a cooling effect on air temperature; the hot summer wind took a moderate trend of variation when frequent rainfall began to take place in the month of May and continued till September (Miah et al. 1981). Michael (1968b) reported that the rainfall and air temperature had the direct influences on the variation of water temperature. Bhuiyan et al. (1997) also observed the seasonal variation of rainfall.
Physical condition

**Water depth**: Maximum depth of water was recorded in August 2001 as 10.90 m whereas the minimum one was recorded in February 2012 as 7.3 m. The average depth of Ramshagar dighi was 9.10 m from May 2011 to March 2014 (Table 1 and Fig. 1). There is a seasonal variation in the depthness of water in water body and rise in water level during monsoon and winter rains has been found. Rahman (1982) stated that ponds should not be shallower than 1.1 m and not deeper than 5 m, and the optimum should be 2 m.

**Fig. 1.** Fluctuation of air temperature (a), relative humidity (b), rainfall (c) and water depth (d) of water of Ramshagar dighi from May 2011 and March 2012.

**Water temperature**: Throughout the study period water temperature was found to fluctuate from the maximum of 32.5°C at the month of June to the minimum of 9.9°C at the month of January (Table 1 and Fig. 2). Average mean of water temperature was determined as 24.68 ± 4.77. Many workers observed similar trends while working on different water bodies (Dwivedi and Pandey 2002). The highest water temperature was recorded in summer (Patra and Azadi 1987) and low water temperature was found in winter months (Das and Bhuiyan 1974) in Bangladesh. The fluctuation in water temperature usually depended on the season, geographic location, sampling time and temperature of effluents entering the stream (Ahipathy and Puttaiah 2006). The standard limit of temperature of water is 20 - 30°C (ECR 1997). All the values were within the standard limit. So, the water bodies are suitable for aquatic life. Therefore, the fluctuation of water temperature in the study area of Ramshagar dighi may be due to the variation of excess CO₂, rainfall and air temperature.

**Redox characteristics**: **pH** of water of Ramshagar dighi was measured monthly where the pH value showed a slightly alkaline in nature during the study period. The pH value of water was found to be fluctuated from a minimum of 6.9 in the month of June, 2011 and maximum of 8.3 in the month of March, 2012 (Table 1 and
During the study period mean of the pH value of water was recorded as 7.67 ± 0.48. Islam et al. (1974) reported the fluctuation of water in Buriganga river, Bangladesh from 7.8 to 6.9 in the months of July and March respectively. pH value of Ramna lake water showed maximum variation as 9.8 in July and 7.5 in March (Islam and Saha 1975). Ahmed et al. 2005 recorded the maximum pH value of water of Meghna river, Bangladesh as 8.00 in the month of September and minimum as 7.5 in the month of May. The pH value in alkaline condition in pond water was supposed to be helpful for proper growth and development of fishes and other aquatic organisms (Nikolsky 1963). Jhingran (1985) has shown that pH range 7-8 is suitable for fish culture as well as most of aquatic organisms. In most raw water sources, pH lies within the range of 6.5 - 8.5 (Ahmed and Rahman 2000). So, these aquatic bodies are suitable for aquatic life. Fluctuation in water pH level found in the present study may due to the change of water temperature and CO₂.

**Free carbon dioxide (CO₂):** Free carbon dioxide is also a determining factor for aquatic ecosystem. Free CO₂ value of the water ranged from 0.00 to 2.2 mg/l in the months of July and December respectively during the period of May 2011 to March 2012. During the study period, mean of free carbon dioxide value of the water was estimated at 0.85 mg/l ± 0.92 (Table 1 and Fig. 2).
Physico-Chemical Properties

Bhuiyan and Nessa (1996) found that free CO$_2$ was absent in May, June and July whereas the highest value (7.35 mg/l) was found in August. Bhuiyan and Nessa (1998a) informed that the free CO$_2$ fluctuated between 0.0 mg/l (January and March) to 15 mg/l (September). So it was found that low free CO$_2$ content during winter and spring and high free CO$_2$ content during autumn. The increase in carbon dioxide level during summer may be due to decay and decomposition of organic matter. This is strengthened by the observations of Joshi et al. (1995) who have observed the addition of drainage was the main causal factor for increase in carbon dioxide in the water bodies.

**Dissolved oxygen (DO):** Dissolve oxygen (mg/l) is the determining factor for all of the aquatic organisms. The dissolved oxygen value of water of the study area ranged from 3.9 mg/l to 5.80 mg/l during the period of May 2011 to March 2012. The maximum DO was recorded 5.80 mg/l in the month of June and the minimum of DO was recorded 3.90 mg/l in the month of August (Table 1 and Fig. 2). Mean value of the DO (mg/l) was remained 4.65 ±0.62 during the period of this study. The dissolved oxygen content of pond water was found maximum during the month of June in Bangladesh (Chowdhury and Mazumder 1987, Khan et al. 1990). Islam et al. (1979) reported minimum dissolved oxygen in September whereas maximum in February. Ali et al. (1989) found high value during winter and low value in summer and a decline in rains. Bhuiyan and Nessa (1996) noted maximum dissolved oxygen during autumn and minimum value in winter. The quantity of DO in water is directly or indirectly dependent on water temperature, partial pressure of air etc. Similar results were observed by Chaurasia and Pandey (2007). The standard DO value of surface water for Bangladesh is 6 or more (ECR 1997). Huq and Alam (2005) mentioned that water with DO value ranging 4 - 6 mg/l is suitable for drinking purpose. In respect of DO, the water body can be used for fish culture and other aquatic organisms. Our results are in close conformity with those reported above. However, more comprehensive works are to be solicited.

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**References**


