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Discovery of potential cholesterol esterase inhibitors using *in silico* docking studies

Thirumalaisamy Sivashanmugam¹, Soundararajan Muthukrishnan², Muthuswamy Umamaheswari¹, Kuppusamy Asokkumar¹, Varadharajan Subhadradevi¹, Puliyaath Jagannath¹ and Arumugam Madeswaran¹

¹Department of Pharmacology, College of Pharmacy, Sri Ramakrishna Institute of Paramedical Sciences, Coimbatore, Tamil Nadu, India; ²Department of Pharmacology, Sankaralingam Bhuvaneshwari College of Pharmacy, Sivakasi, Tamil Nadu, India.

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Abstract

New drug discovery is considered broadly in terms of two kinds of investigational activities such as exploration and exploitation. This study deals with the evaluation of the cholesterol esterase inhibitory activity of flavonoids apigenin, biochanin, curcumin, diosmetin, epipervilline, glycitein, okanin, rhamnazin and tangeritin using *in silico* docking studies. *In silico* docking studies were carried out using AutoDock 4.2, based on the Lamarckian genetic algorithm principle. The results showed that all the selected flavonoids showed binding energy ranging between -7.1 kcal/mol to -5.6 kcal/mol when compared with that of the standard compound gallic acid (-4.1 kcal/mol). Intermolecular energy (-9.1 kcal/mol to -7.1 kcal/mol) and inhibition constant (6.5 μ M to 73.2 μ M) of the ligands also coincide with the binding energy. All the selected flavonoids contributed cholesterol esterase inhibitory activity, these molecular docking analyses could lead to the further development of potent cholesterol esterase inhibitors for the treatment of obesity.

Introduction

Obesity is becoming one of the greatest threats to global health in this millennium, with more than 1 billion overweight adults and of those, at least 300 million are clinically obese. The obesity market reached US\$ 3.7 billion by 2008 and has been predicted to reach US\$ 6.1 billion by 2015. The mushrooming market for these drugs and the vast sum of money at stake guarantee that research in this therapeutic area will not slow down within the predictable (Birari and Bhutani, 2007). Obesity is resulting from an energy imbalance caused by increased ratio of caloric intake to energy expenditure and the excess body fats thus formed are stored in body. Obesity is also known to be risk factor for the development of metabolic disorders, dyslipidemia and

atherosclerosis and type-2 diabetes. In recent years, there has been a great increase in the use of herbal medicines for the treatment of obesity (Ramgopal et al., 2010). Overweight and obesity are the fifth leading risk for global deaths. At least 2.8 million adults die each year as a result of being overweight or obese (Barnes et al., 2007)

Pancreatic cholesterol esterase (CEase) is secreted from vertebrate pancreas into the intestinal track and activated by primary bile salts. It is present in few mammals including humans (John et al., 2010). CEase belongs to α/β -hydrolase fold family. The active site includes both the catalytic triad (Ser-194, Asp-320, His-435) and oxyanion hole (Gly-107, Ala-108, Ala-195) residues (John et al., 2010). CEase catalyzes the



hydrolysis of sterol esters into their component sterols and fatty acids. It is responsible for the hydrolysis of various substrates including dietary cholesterol esters, fat-soluble vitamins, triglycerides, and phospholipids. Flavonoids are widespread phytochemical constituents present in plants and they contribute the flavor and color of fruits and vegetables. They are 15 carbons of 2 phenolic rings connected by a 3-carbon unit and grouped according to presence of various functional groups on the rings and the degree of ring saturation (Whittern et al., 1984). They are usually attached with sugars moiety to increase their water-solubility. Most of the flavonoids are known to possess various pharmacological activities, such as anti-inflammatory activity (Funakoshi et al., 2011), neuroprotective effects (Datla et al., 2001), antibacterial (Martini et al., 2001; Jin et al., 2012), anti-oxidant (Pande, 2004), anti-proliferative (Harinantenaina et al., 2010), cytostatic effect (Androtsopoulos et al., 2012), and antiviral (Da et al., 2010).

Nowadays, the use of computers to predict the binding of libraries of small molecules to known target structures is an increasingly important component of the drug discovery process (Schoichet, 2004; Koppen, 2009). There is a wide range of software packages available for the conduct of molecular docking simulations like, AutoDock and DOCK, GOLD, FlexX and ICM (Collignon et al., 2011). AutoDock 4.2 is the most recent version which has been widely used for virtual screening, due to its enhanced docking speed (Schames et al., 2004). Its default search function is based on Lamarckian Genetic Algorithm (LGA), a hybrid genetic algorithm with local optimization that uses a parameterized free-energy scoring function to

estimate the binding energy. Each docking is comprised of multiple independent executions of LGA and a potential way to increase its performance is to parallelize the aspects for execution (Cosconati et al., 2010).

The stereochemistry of flavonoids binding on CEase has not been characterized. In this study, the search of flavonoids in the molecular basis for binding to active site of CEase is revealed by computer aided docking analysis.

Materials and Methods

Softwares required

Python 2.7-language was downloaded from www.python.com, Cygwin (a data storage) `c:\program` and Python 2.5 were simultaneously downloaded from www.cygwin.com, Molecular graphics laboratory (MGL) tools and AutoDock 4.2 was downloaded from www.scripps.edu, Discovery studio visualizer 2.5.5 was downloaded from www.accelerys.com, Molecular orbital package (MOPAC), Chems sketch was downloaded from www.acdlabs.com. Online smiles translatory notation was carried out using cactus.nci.nih.gov/translate/.

Coordinate file preparation

An extended PDB format, termed as PDBQT file was used for coordinate files which includes atomic partial charges. AutoDock Tools was used for creating PDBQT files from traditional PDB files (Khodade et al., 2007). Crystal structure of CEase (PDB ID: 1F6W) was



Figure 1: Cholesterol esterase enzyme from RCSB (1F6W)

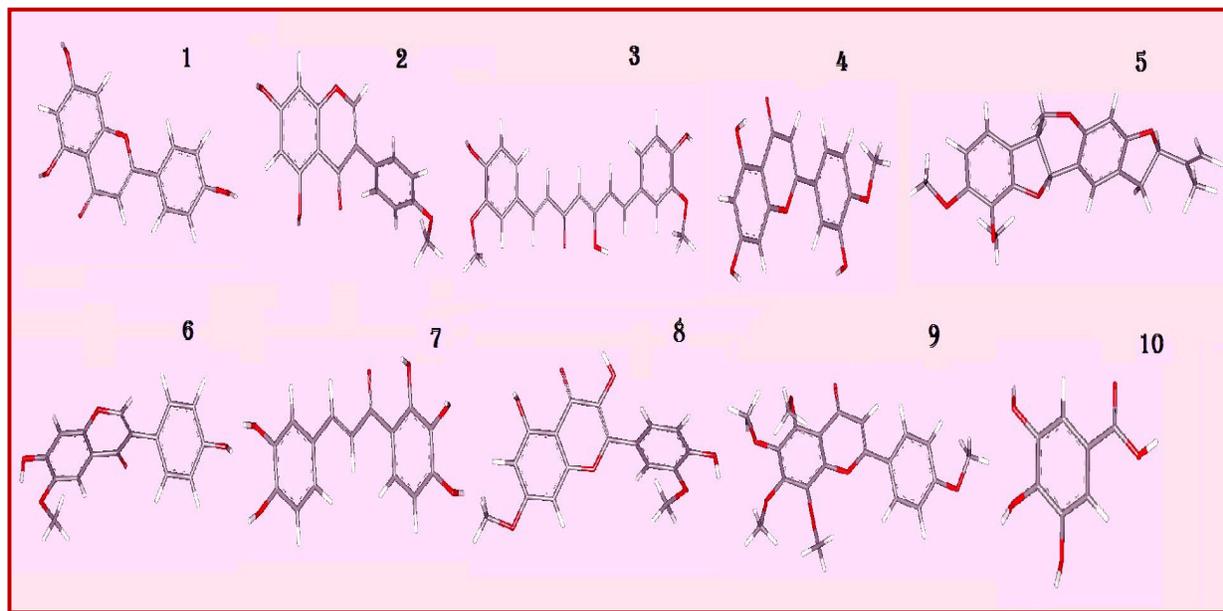


Figure 2: The optimized ligand molecules (1. Apigenin, 2. Biochanin, 3. Curcumin, 4. Diosmetin, 5. Epipervilline, 6. Glycitein, 7. Okanin, 8. Rhamnazin, 9. Tangeritin and 10. Gallicacid)

downloaded from the RCSB protein data bank (Figure 1).

In Figure 2, the flavonoid ligands like apigenin, biochanin, curcumin, diosmetin, epipervilline, glycitein, okanin, rhamnazin, tangeritin and the standard gallicacid were built using Chemskech and optimized using "Prepare Ligands" in the AutoDock 4.2 for docking studies. The optimized ligand molecules were docked into refined CEase model using "LigandFit" in the AutoDock 4.2 (Goodsell et al., 1996).

AutoGrid calculation

Rapid energy evaluation was achieved by precalculating atomic affinity potentials for each atom in the ligand molecule. In the AutoGrid procedure, the target enzyme was embedded on a three dimensional grid point (Morris et al., 1998). The energy of interaction of each atom in the ligand was encountered.

AutoDock calculation

Docking can be carried out by various methods. But, the most efficient method is Lamarckian genetic algorithm. AutoDock was run several times to get various docked conformations, and used to analyze the predicted docking energy. The binding sites for these molecules were selected based on the ligand-binding pocket of the templates (Chang et al., 2010).

Analysis using AutoDock Tools

AutoDock Tools provide various methods to analyze the results of docking simulations such as, conformational similarity, visualizing the binding site and its energy and other parameters like intermolecular energy and inhibition constant. For each ligand, ten best poses

were generated and scored using AutoDock 4.2 scoring functions (Park et al., 2006).

Results and Discussion

The docking poses were ranked according to their docking scores and both the ranked list of docked ligands and their corresponding binding poses (Zhang et al., 2008). In Figure 3, docked pose of CEase enzyme with epipervilline ligand clearly demonstrated the binding positions of the ligand with the enzyme. Binding energy of the individual compound were calculated using the following formula:

$$\text{Binding energy} = A + B + C - D$$

Where, A denotes final intermolecular energy + Vander valls energy (vdW) + hydrogen bonds + desolvation energy + electrostatic energy (kcal/mol), B denotes final total internal energy (kcal/mol), C denotes torsional free energy (kcal/mol), D denotes unbound system's energy (kcal/mol)

In the Figure 3, the binding sites of the most active flavonoid Epipervilline was clearly showed and it was found to be, GLY-106, GLY-107, ALA-108, TYR-125, GLU-193, SER-194, TRP-227, VAL-285, LEU-282, PHE-324, LEU-392, PHE-393, HIS-435 and ILE-439.

In Figure 4, the potential binding sites of the standard was clearly showed and it was found that, ALA-117, ASN-118, PHE-119, ASN-121, ASN-122, TYR-123, LEU-124, LYS-445. This proves that the effective binding sites are present in the selected flavonoid epipervilline when compared with the standard gallic acid. It proves that the ability of inhibiting the CEase enzyme by the selected flavonoid.

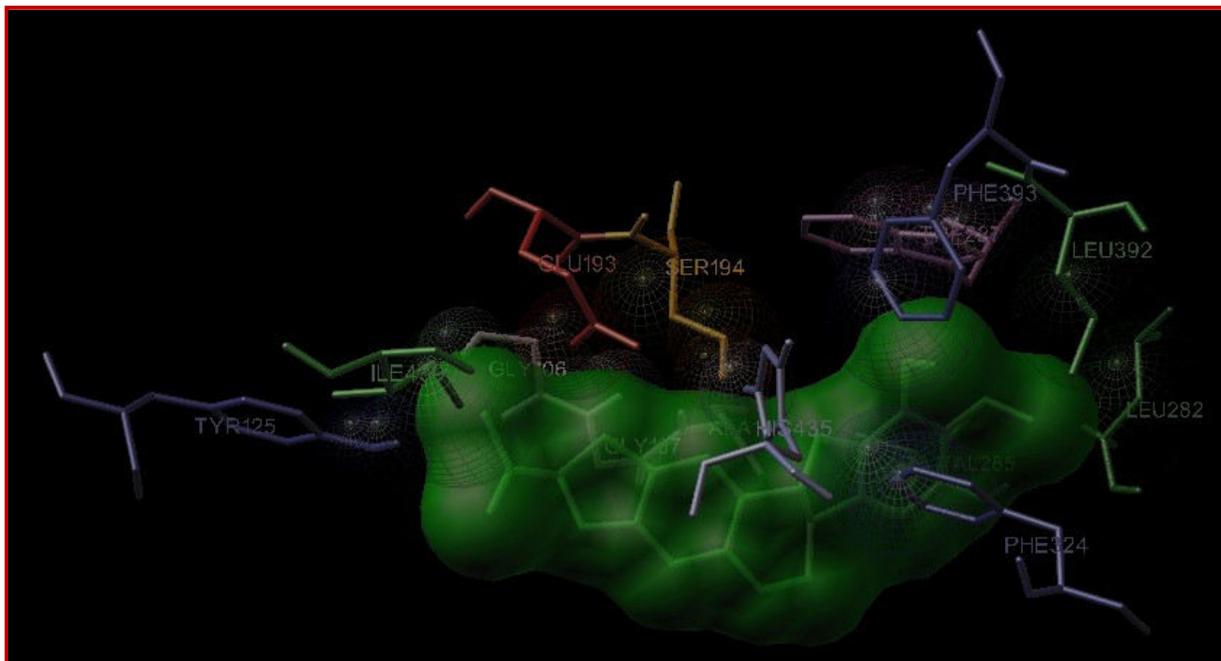


Figure 3: Docked pose of cholesterol esterase enzyme (1F6W) with Epipervilline

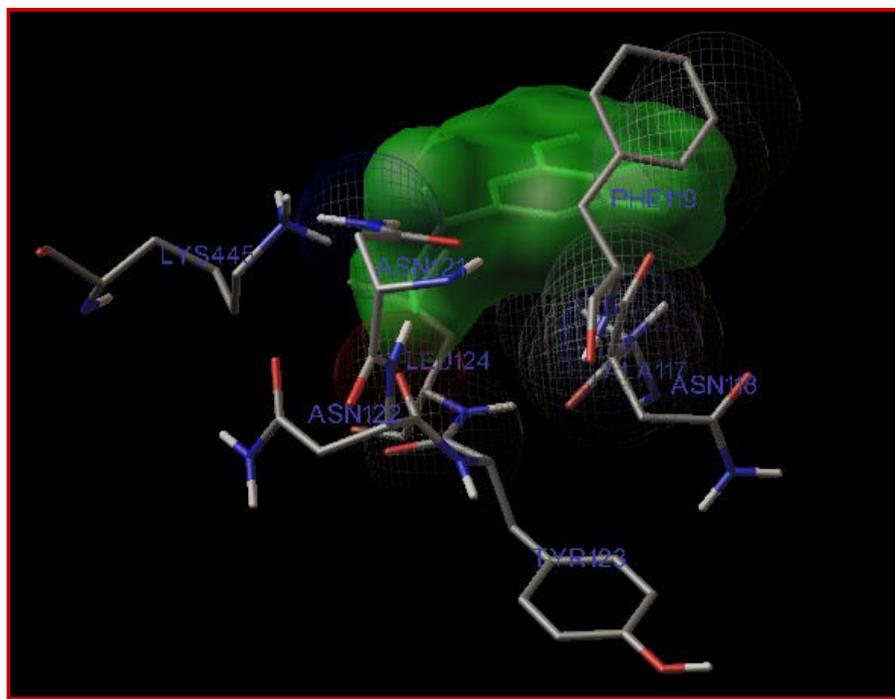


Figure 4: Docked pose of cholesterol esterase enzyme (1F6W) with gallic acid

Analysis of the receptor/ligand complex models generated after successful docking of the flavonoids was based on the parameters such as, hydrogen bond interactions, $\pi - \pi$ interactions, binding energy, RMSD of active site residues and orientation of the docked compound within the active site. As a general rule, in most of the potent antigout compounds, both hydrogen bond and $\pi - \pi$ hydrophobic interactions between the

compound and the active sites of the receptor have been found to be responsible for mediating the biological activity (Umamaheswari et al., 2011).

As shown in Table I, flavonoids showed binding energy ranging between -7.1 to -5.6 kcal/mol. All the selected flavonoids had lesser binding energy when compared to the standard gallic acid (-4.1 kcal/mol). This proves

Table I										
Binding energies of the compounds based on their rank										
Compound	Binding energies of the compounds based on their rank (kcal/mol ⁻¹)									
	1	2	3	4	5	6	7	8	9	10
Apigenin	-5.9	-5.9	-5.7	-5.7	-5.7	-5.7	-5.7	-5.7	-5.7	-5.7
Biochanin	-6.5	-6.5	-6.5	-6.5	-6.5	-6.5	-6.5	-6.5	-6.5	-6.4
Curcumin	-6.1	-5.1	-5.5	-5.0	-5.0	-5.0	-5.3	-5.1	-5	-4.8
Diosmetin	-6.0	-6.0	-6	-6.0	-6.0	-5.9	-5.9	-5.9	-5.8	-5.8
Epipervilline	-7.1	-7.1	-7.1	-7.1	-7.1	-7.1	-7.0	-7.0	-6.6	-6.5
Glycitein	-6.1	-5.5	-5.5	-5.5	-5.4	-5.4	-5.4	-5.3	-5.3	-5.2
Okanin	-6.4	-5.4	-5.3	-5.0	-5.0	-4.6	-4.5	-5.1	-4.9	-4.8
Rhamnazin	-5.6	-5.6	-5.2	-5.0	-4.9	-4.6	-4.4	-4.8	-4.5	-4.8
Tangeritin	-6.1	-5.9	-5.6	-5.6	-5.4	-5.3	-5.5	-5.3	-5.1	-5.1
Gallic acid	-4.1	-4.1	-4.1	-4.1	-4.1	-4.1	-4.1	-4.1	-4.1	-4.0

Table II										
Inhibition constant of the compounds based on their rank										
Compounds	Inhibition constant of the compounds based on their rank (μM , mM ^a)									
	1	2	3	4	5	6	7	8	9	10
Apigenin	47.7	48.8	62.7	66.1	67.1	68.0	65.0	65.2	66.6	66.6
Biochanin	16.9	17.0	17.1	17.1	17.1	17.1	17.2	17.2	17.2	19.2
Curcumin	31.4	193.0	99.1	211.5	224.2	225.8	122.9	179.9	217.5	331.7
Diosmetin	37.3	39.4	39.9	40.4	42.6	44.0	45.3	45.4	53.1	54.0
Epipervilline	6.5	6.6	6.6	6.8	6.8	6.9	7.2	7.2	15.1	18.2
Glycitein	33.8	93.0	100.3	94.7	112.7	113.2	113.6	138.6	130.6	149.1
Okanin	20.3	116.9	130.7	204.3	212.7	441.5	49.3	185.9	257.8	332.2
Rhamnazin	73.2	80.6	159.3	227.0	263.0	411.4	600.2	283.0	545.8	318.8
Tangeritin	33.2	50.6	75.5	79.7	114.1	125.9	97.9	141.5	75.2	172.8
Gallic acid	357.8	378.0	541.9	565.3	568.4	930.3	1.0 ^a	511.0	656.3	970.3

that flavonoids consist of potential CEase inhibitory binding sites when compared to the standard. In addition, two other parameters like inhibition constant (K_i) and intermolecular energy were also determined. As shown in Table II, flavonoids showed inhibition constant ranging from 6.5 to 73.2 μM . All the selected compounds had lesser inhibition constant when compared to the standard (357.8 μM). Inhibition constant is directly proportional to binding energy. We found a decrease in inhibition constant of all the selected flavonoids with a simultaneous decrease in the binding energy. Thus, the cholesterol esterase inhibitory activity of the flavonoids were found to be higher compared to gallic acid.

As shown in Table III, flavonoids showed intermolecular energy ranging between -9.1 to -7.1 kcal/mol which was lesser when compared to the standard (-6.2

kcal/mol). Intermolecular energy is also directly proportional to binding energy. We found a decrease in intermolecular energy of all the selected compounds with a simultaneous decrease in the binding energy. This result further proved the CEase inhibitory activity of all the selected flavonoids.

Based on the docking studies, the CEase inhibitory activity of the selected compounds was found to be decreased in the order of epipervilline, biochanin, okanin, curcumin, tangeritin, glycitein, diosmetin, apigenin, rhamnazin and gallic acid. On the basis of the above study, all the selected compounds possess potential CEase inhibitory binding sites when compared to that of the standard. This may be attributed due to the differences in the position of the functional groups in the compounds. Flavonoids consist of benzopyran ring in its basic nucleus which could be

Table III

Intermolecular energies of the compounds based on their rank

Compound	Intermolecular energies of the compounds based on their rank (kcal/mol)									
	1	2	3	4	5	6	7	8	9	10
Apigenin	-7.1	-7.1	-6.9	-6.9	-6.9	-6.9	-6.9	-6.9	-6.9	-6.9
Biochanin	-7.7	-7.7	-7.7	-7.7	-7.7	-7.7	-7.7	-7.7	-7.7	-7.6
Curcumin	-9.1	-8.1	-8.5	-8	-8.0	-8.0	-8.3	-8.1	-8.0	-6.9
Diosmetin	-7.5	-7.5	-7.5	-7.5	-7.5	-7.4	-7.4	-7.4	-7.3	-7.3
Epipervilline	-8.0	-8.0	-8.0	-8.0	-7.9	-7.9	-7.9	-7.9	-7.5	-7.4
Glycitein	-7.3	-6.7	-6.7	-6.7	-6.6	-6.6	-6.6	-6.5	-6.5	-6.3
Okanin	-8.8	-7.8	-7.7	-7.4	-7.4	-7.0	-6.9	-7.5	-7.3	-7.1
Rhamnazin	-7.4	-7.4	-7.0	-6.8	-6.7	-6.4	-6.2	-6.6	-6.2	-6.6
Tangeritin	-7.9	-7.7	-7.4	-7.4	-7.2	-7.1	-7.3	-7.0	-6.9	-6.9
Gallic acid	-6.2	-6.2	-6.0	-5.9	-5.9	-5.6	-5.6	-6.0	-5.8	-5.6

responsible for the CEase inhibitory activity.

These results clearly indicate that flavonoids especially, Epipervilline possess potential CEase inhibitory binding sites and further investigations on the above compounds are necessary to develop potential chemical entities for the prevention and treatment of obesity and related disorders.

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Author Info

Thirumalaisamy Sivashanmugam (Principal contact)
e-mail: pharmsiva@gmail.com